

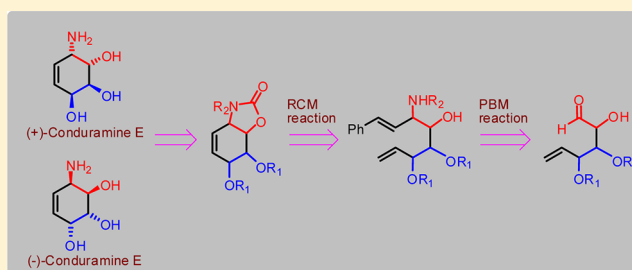
A Chiron Approach to Aminocyclitols by Petasis-Borono-Mannich Reaction: Formal Synthesis of (+)-Conduramine E and (–)-Conduramine E

Partha Ghosal and Arun K. Shaw*

Division of Medicinal and Process Chemistry, CSIR-Central Drug Research Institute (CDRI), Lucknow 226 001, India

Supporting Information

ABSTRACT: A chiron approach to a stereoselective route for the synthesis of aminocyclitols from carbohydrates is described. The formal synthesis of (+)-conduramine E and (–)-conduramine E was achieved by utilizing this strategy. The key features of the synthetic strategy include one-pot three-component Petasis-Borono-Mannich reaction to introduce the *syn*- β -amino alcohol functionality of conduramine E and ring-closing metathesis to construct its carbocyclic core. The present synthetic approach paves the way for stereoselective synthesis of several conduramines starting from carbohydrates.



Aminocyclohexenetriols formally called conduramines are the amino derivatives of conduritols, in which an amino functionality is present in place of one of the hydroxyl groups (Figure 1). These densely functionalized aminocyclitols and

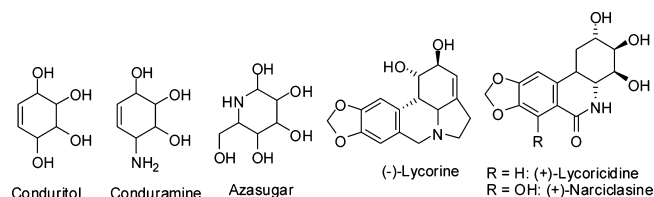


Figure 1. Structure of some important aminocyclitols.

structurally related compounds form the basic skeleton of pseudo oligosaccharides and several complex aminoglycoside antibiotics.^{1–3} Such units are also the synthetic precursors of pharmacologically important molecules such as Pancratistatin,⁴ (+)-lycoridine,⁵ (+)-narciclasine,⁶ and (–)-lycorine,⁷ with important bioactivities such as antibacterial, antihypertensive, platelet-inhibiting, and cytotoxic (Figure 1). In addition, utilization of aminocyclitols for the syntheses of azasugars,⁸ aminosugars,⁹ sphingosines,¹⁰ and narcissus alkaloids¹¹ amplified their importance as synthetic building blocks. Therefore, several elegant approaches to the synthesis of racemic as well as optically pure aminocyclitols and their structural variants have been disclosed to date.^{12–27} Very recently Norsikian et al. disclosed a novel approach to conduramines starting from carbohydrate via an unprecedented intramolecular Petasis-Borono-Mannich reaction with an exclusive *anti* stereoselectivity for the newly created β -amino alcohol motif.²⁸

The interesting structural features associated with biological activities and also as a result of our involvement toward the synthesis of carbohydrate-based chiral building blocks

(CBBs)²⁹ and their exploitation in syntheses of natural products or natural-product-like molecules of significant biological importance^{30–36} prompted us to develop an efficient methodology for synthesis of enantiomerically pure aminocyclohexenetriols by a chiron approach. Herein, we wish to disclose a short and efficient formal synthesis of both enantiomers of conduramine E (Figure 2) from carbohydrates involving Petasis-Borono-Mannich (PBM) chemistry^{37–41} and ring-closing metathesis.

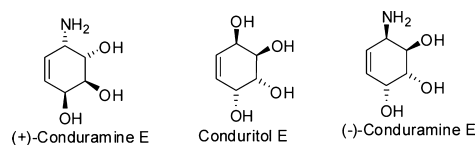


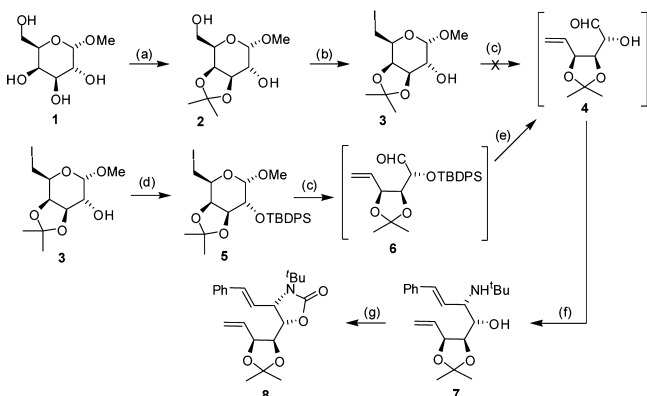
Figure 2. Structures of condurititol E, (+)-conduramine E, and (–)-conduramine E.

The synthesis of (+)-conduramine E was commenced with the selective acetonide protection of hydroxyls at C3 and C4 in methyl- α -D-galactopyranoside **1** to give 3,4-*O*-isopropylidene-methyl- α -D-galactopyranoside **2** (Scheme 1). Iodination of the primary alcohol was carried out by treating the diol **2** under Garegg and Samuelsson conditions to obtain the iodide **3**, which on treatment with Zn dust in the presence of a catalytic amount cyanocobalamin⁴² (vitamin B₁₂) gave the required hydroxy aldehyde **4**, but unfortunately the α -hydroxy aldehyde **4** was unstable (Scheme 1).

At this stage, we modified our synthetic strategy by protecting the remaining hydroxyl group at C2 in **3** as its

Received: April 27, 2012

Published: August 17, 2012

Scheme 1. Synthesis of Oxazolidinone 8^a

^aReagent and conditions: (a) I₂, acetone, 4 h, 70%; (b) I₂, PPh₃, imidazole, toluene, 80 °C, 3 h, 82%; (c) Zn, NH₄Cl, vit. B₁₂, MeOH, 5 min; (d) TBDPSCl, imidazole, DCM, 6 h, 85%; (e) TBAF, EtOH, 30 min; (f) *trans*-2-phenylvinyl boronic acid, ^tBuNH₂, 80 °C, 24 h; (g) (Boc)₂O, Et₃N, DMAP, THF, rt, 12 h, 15% over three steps.

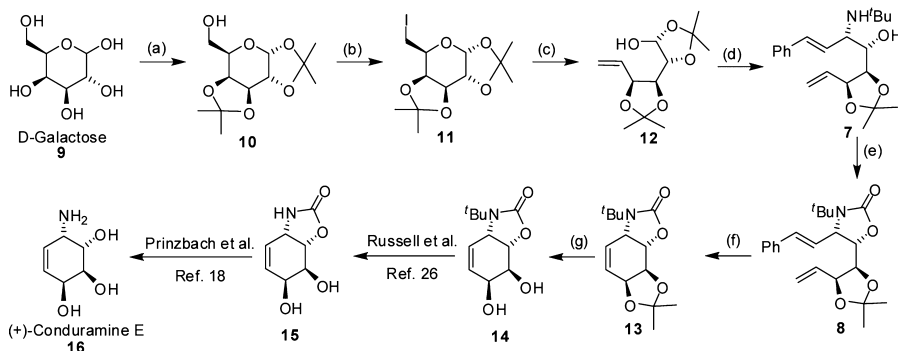
silyl ether. Thus, the iodide 3 was protected as its silyl ether 5, which on Zn-dust-mediated ring opening in the presence of catalytic amount cyanocobalamin afforded the stereochemically pure silyl protected hydroxy aldehyde 6. It was then subjected to one-pot desilylation by TBAF followed by PBM reaction of the resulting aldehyde 4 with *trans*-2-phenylvinyl boronic acid and *tert*-butylamine under refluxing condition to obtain the unsaturated amino alcohol 7 (Scheme 1). Since it is well documented in the literature that the chiral α -hydroxy aldehyde furnishes the corresponding *erythro*-1,2 amino alcohol as a single diastereomer,⁴¹ herein the *erythro*-1,2 amino alcohol 7 was also obtained exclusively as a single diastereoisomer from the aldehyde 6. Unfortunately, we could not isolate 7 in its pure form by column chromatographic purification of the crude reaction product (TLC). At this stage, the amino group in 7 was protected by treating it with (Boc)₂O in THF in the presence of Et₃N and DMAP. Here, the Boc protection of the amine 7 with simultaneous oxazolidinone ring formation furnished the oxazolidinone 8 with low yield (15%) in three steps (Scheme 1).

The low yield of 8 was presumably attributed to the instability of the α -hydroxy aldehyde intermediate 4, and therefore to evade this problem we decided to start our

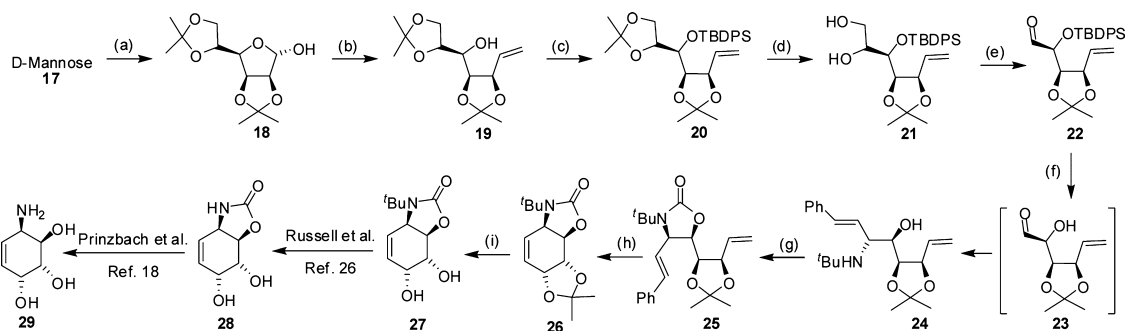
synthetic strategy from D-galactose 9. Thus, the 1,2,3,4-*O*-isopropylidene- α -D-galactopyranoside⁴³ 10 prepared from 9 on iodination furnished iodide 11 in 86% yield. Its reaction with Zn dust in the presence of a catalytic amount of cyanocobalamin smoothly afforded the hemiacetal 12, which was then subjected to PBM reaction with *trans*-2-phenylvinyl boronic acid and *tert*-butylamine to afford exclusively the *erythro*-1,2 amino alcohol 7. Its treatment with (Boc)₂O in THF in the presence of Et₃N/DMAP furnished the oxazolidinone 8 in 40% yield over two steps (Scheme 2). The cyclization of 8 was achieved by using ring-closing metathesis reaction in the presence of Grubbs second generation catalyst to afford conduramine core 13. Cleavage of the acetonide protection in 13 was done by treating it with TFA to obtain diol 14, a known intermediate reported by Russell et al. during (\pm)-conduramine E synthesis.²⁶ Its spectral data was in close agreement with the data reported by Russell et al. Now, the synthesis of (+)-conduramine E 16 could be completed by TFA-mediated ^tBu deprotection²⁶ of 14 followed by basic hydrolysis of the oxazolidinone ring¹⁸ in 15 by adopting the procedure reported by Prinzbach et al. (Scheme 2).

We were also interested in synthesis of (–)-conduramine E (29), the enantiomer of (+)-conduramine E, involving the PBM reaction on masked α -hydroxy aldehyde of the type 22. We achieved our goal by adopting the synthetic strategy shown in Scheme 3. Thus, the 2,3,5,6-*O*-isopropylidene- α -D-mannofuranose hemiacetal 18 prepared from D-mannose 17 by the standard literature procedure (I₂/acetone)⁴⁴ was subjected to undergo Wittig methylenation on anomeric carbon to furnish the alkene 19. The free hydroxyl group in 19 was protected with TBDPSCl to afford silyl ether 20. The selective deprotection of terminal acetonide in 20 was done by treating it with 80% aqueous AcOH to obtain the diol 21 (Scheme 3). Its oxidative cleavage with NaIO₄ afforded an aldehyde 22, which was desilylated with TBAF followed by PBM reaction of the resulting α -hydroxy aldehyde 23 with *trans*-2-phenylvinyl boronic acid and *tert*-butylamine under refluxing condition to afford the desired amine 24 in a one-pot fashion. Its protection with (Boc)₂O in THF in the presence of Et₃N/DMAP furnished the oxazolidinone 25 (Scheme 3).

The diene 25 was the enantiomer of diene 8. The NMR spectra of these two enantiomers (8 and 25) were identical, and their optical rotation values were close in magnitude but opposite in sign $\{[\alpha]_D^{28} = -55.3$ (*c* 0.77, CHCl₃) for 8; $[\alpha]_D^{26}$

Scheme 2. Formal Synthesis of (+)-Conduramine E 16^a

^aReagent and conditions: (a) I₂, acetone, 24 h, 80%; (b) I₂, PPh₃, imidazole, toluene, 80 °C, 3 h, 86%; (c) Zn, NH₄Cl, Vit. B₁₂, MeOH, 15 min, 90%; (d) *trans*-2-phenylvinyl boronic acid, ^tBuNH₂, EtOH, rt, 24 h; (e) (Boc)₂O, Et₃N, DMAP, THF, rt, 12 h, 40% over two steps; (f) Grubbs second generation catalyst (10 mol %), DCM, reflux, 24 h, 51%; (g) TFA, DCM, 2 h then H₂O, 1 h, 85%.

Scheme 3. Formal Synthesis of (–)-Conduramine E 29^a

^aReagent and conditions: (a) I₂, acetone, rt, 24 h, 80%; (b) Ph₃PCH₃Br, ^tBuOK, THF, –20 °C, 4 h, 85%; (c) TBDPSCl, imidazole, DMF, rt, 24 h, 87%; (d) 80% AcOH, rt, 6 h, 75%; (e) NaIO₄, THF/H₂O (9:1), 0 °C to rt, 1.5 h; (f) TBAF, EtOH, 30 min then *trans*-2-phenylvinyl boronic acid, ^tBuNH₂, 80 °C, 24 h; (g) (Boc)₂O, Et₃N/DMAP (3:1), 0 °C to rt, 12 h, 21% from 21; (h) Grubbs second generation catalyst (10 mol %), DCM, reflux, 24 h, 48%; (i) TFA, DCM, 2 h then H₂O, 91%.

= +66.0 (c 0.24, CHCl₃) for 25}. Ring-closing metathesis of diene 25 in the presence of Grubbs second generation catalyst under refluxing condition furnished the required carbocyclic core of (–)-conduramine E 26 (Scheme 3). Finally, the isopropylidene protection in 26 was easily cleaved by treating it with TFA to furnish the diol 27 in 91% yield, from which the (–)-conduramine E 29 could be obtained via 28 by following the procedure reported by Russell et al.²⁶ Here, it was worth mentioning that 26 and 27 were the enantiomers of 13 and 14, respectively. The values for their optical rotations were close in magnitude but opposite in sign {[α]_D²⁸ = +124.2 (c 0.125, CHCl₃) for 13 and [α]_D²⁶ = –134.0 (c 0.112, CHCl₃) for 26}{[α]_D²⁸ = +78.5 (c 0.23, CHCl₃) for 14 and [α]_D²⁶ = –73.7 (c 0.16, CHCl₃) for 27}.

Here, it is worth mentioning that in our present approach the *syn*-β-amino alcohol motif of conduramine E was installed via an intermolecular Petasis-Borono-Mannich reaction, whereas Norsikian et al. disclosed a novel approach to synthesize *anti*-β-amino alcohol motif of conduramine C-4 and *ent*-conduramine A-1 by utilizing intramolecular Petasis-Borono-Mannich reaction.²⁸ Carbohydrates were utilized as a starting material in both the cases.

In summary, we developed a convenient chiron approach to the formal synthesis of both the enantiomers of conduramine E. The commercially available monosaccharides were utilized as starting material to complete the synthesis of key oxazolidinone fused carbocyclic intermediates (13 and 26) involving highly diastereoselective Petasis-Borono-Mannich (PBM) and ring-closing metathesis (RCM) reactions. Several aminocytitols could also be synthesized by using this flexible approach starting from carbohydrates.

EXPERIMENTAL SECTION

Compound 2. To a stirred suspension of methyl-α-D-galactopyranoside 1 (500 mg, 2.57 mmol) in dry acetone (10 mL) was added I₂ (130 mg, 0.5 mmol) at room temperature, and the reaction was allowed to stir at same temperature for 4 h. After completion of the reaction, excess iodine was quenched with saturated aqueous solution of Na₂S₂O₃ and the resulting mixture was concentrated. The reaction mixture was then extracted with EtOAc (3 × 20 mL) and the combined organic layer was dried over Na₂SO₄ and concentrated under reduced pressure to a residue that on purification by column chromatography gave 2,3-*O*-isopropylidene-methyl-α-D-galactopyranoside 2 (420 mg, 1.8 mmol, 70%).

Eluent for column chromatography: EtOAc/hexane (2/1, v/v); [α]_D²⁷ = +156.7 (c 0.8, CHCl₃); R_f = 0.25 (2/1, EtOAc/hexane); IR

(neat, cm⁻¹) 3336, 2864, 1639, 1222, 1076, 770; ¹H NMR (300 MHz, CDCl₃) δ 1.28 (s, 3H), 1.44 (s, 3H), 2.83 (s, 1H, OH), 3.11 (m, 1H, OH), 3.39 (s, 3H), 3.72–3.76 (m, 2H), 3.82–3.89 (m, 1H), 3.97–3.99 (m, 3H), 4.17–4.18 (m, 2H), 4.71 (d, J = 3.6 Hz, 1H); ¹³C NMR (50 MHz, CDCl₃) δ 26.2, 28.0, 55.8, 62.7, 68.4, 69.9, 74.1, 76.6, 99.1, 110.0; HRMS (ESI TOF (+)) *m/z* [M + H]⁺ calcd for C₁₀H₁₉O₆ 235.1182, found 235.1174.

Compound 3. 2,3-*O*-Isopropylidene-methyl-α-D-galactopyranoside 2 (500 mg, 2.14 mmol), PPh₃ (840 mg, 3.2 mmol) and imidazole (435 mg, 6.4 mmol) were taken in a round-bottom flask in dry toluene (15 mL). The reaction mixture was stirred at room temperature with I₂ (815 mg, 3.2 mmol). The dark brown reaction mixture was heated at 80 °C for 3 h. After completion of the reaction, the reaction mixture was quenched with MeOH (2 mL), and the whole mixture was then concentrated to a residue. Saturated aqueous solution of Na₂S₂O₃ was added to it. The entire solution was stirred until the reaction mixture became colorless. Afterward, it was extracted with DCM (2 × 20 mL), and the combined organic layer was dried over Na₂SO₄ and concentrated under reduced pressure to a residue which on column chromatographic purification afforded iodide 3 (600 mg, 1.75 mmol, 82%).

Eluent for column chromatography: EtOAc/hexane (2/5, v/v); [α]_D²⁸ = +130.3 (c 0.74, CHCl₃); R_f = 0.32 (1/3, EtOAc/hexane); IR (neat, cm⁻¹) 3133, 3077, 2542, 1384, 1215, 1021, 781; ¹H NMR (300 MHz, CDCl₃ + CD₃OD) δ 1.35 (s, 3H), 1.50 (s, 3H), 3.33–3.38 (m, 2H), 3.48–3.49 (m, 3H), 3.73–3.74 (m, 1H), 4.127–4.134 (m, 1H), 4.19–4.23 (m, 1H), 4.33–4.40 (m, 3H), 4.71 (m, 1H); ¹³C NMR (50 MHz, CDCl₃ + CD₃OD) δ 6.1, 29.7, 31.5, 59.4, 72.6, 73.3, 77.7, 80.2, 103.5, 113.3; HRMS (ESI TOF (+)) *m/z* [M + H]⁺ calcd for C₁₀H₁₈IO₅ 345.0199, found 345.0191.

General Procedure for Preparation of Aldehyde 4. To a magnetically stirred suspension of Zn dust (680 mg, 10.4 mmol) and NH₄Cl (560 mg, 10.38 mmol) in dry methanol (20 mL) was added cyanocobalamin (7 mg, 0.005 mmol), and the resulting mixture was allowed to stir for 15 min. Afterward, a solution of 3 (150 mg, 0.44 mmol) in dry methanol (5 mL) was added to it, and the resulting solution was further stirred for 5 min. The reaction mixture was filtered through a Celite bed, and the filtrate was evaporated under reduced pressure. The residue obtained was dissolved in ethyl acetate (25 mL) and washed with a mixture of brine and water (1:1 v/v, 10 mL each). The organic layer was dried over Na₂SO₄ and concentrated to give the crude aldehyde 4, which unfortunately decomposed rapidly.

Compound 5. To a stirred solution of compound 3 (345 mg, 1 mmol) in dry DCM (5 mL) were added imidazole (135 mg, 2 mmol) and TBDPSCl (412 mg, 0.39 mmol, 1.5 mmol), and the reaction mixture was allowed to stir for another 6 h. Water (10 mL) was added to the reaction mixture and extracted with DCM (2 × 10 mL). The combined organic layer was evaporated under reduced pressure to give a residue that on purification by column chromatography gave 5 as clear oil (495 mg, 0.85 mmol, 85%).

Eluent for column chromatography: EtOAc/hexane (1/19, v/v); $[\alpha]_D^{28} = +84.2$ (c 1.23, CHCl_3); $R_f = 0.54$ (1/6, EtOAc/hexane); IR (neat, cm^{-1}) 3017, 2361, 1372, 1215, 763; ^1H NMR (300 MHz, CDCl_3) δ 1.12 (s, 9H), 1.24 (s, 3H), 1.33 (s, 3H), 3.23–3.34 (m, 5H), 3.73–3.77 (m, 1H), 4.08–4.13 (m, 1H), 4.26–4.37 (m, 3H), 7.36–7.44 (m, 6H), 7.71–7.77 (m, 4H); ^{13}C NMR (50 MHz, CDCl_3) δ 3.2, 19.6, 26.7, 27.3, 28.2, 55.6, 86.4, 72.2, 74.3, 77.4, 128.0, 130.1, 133.1, 134.7, 136.1, 136.5; HRMS (ESI TOF (+)) calcd for $\text{C}_{26}\text{H}_{35}\text{IO}_3\text{SiNa}$ $[\text{M} + \text{Na}]^+$ 605.1196, measured 605.1208.

Compound 8. To a magnetically stirred suspension of Zn dust (1.56 mg, 23.9 mmol) and NH_4Cl (1.3 g, 23.9 mmol) in dry methanol (30 mL) was added cyanocobalamin (16 mg, 0.012 mmol), and stirring was continued for another 15 min. After that, a solution of **5** (580 mg, 0.1 mmol) in dry methanol (5 mL) was added, and the resulting solution was further stirred for 5 min. The reaction mixture was filtered through a Celite bed. The filtrate was evaporated under reduced pressure. The residue obtained was dissolved in ethyl acetate (25 mL) and washed with a mixture of brine and water (1:1 v/v, 10 mL each). The organic layer was dried over Na_2SO_4 and concentrated to give the crude aldehyde **6** (430 mg), which was used immediately for the next step.

To a stirred solution of aldehyde **6** in EtOH (10 mL) was added TBAF (1 mL, 1 M solution in THF), and the reaction mixture was left stirring at room temperature. After 30 min, *trans*-2-phenylvinyl boronic acid (149 mg, 1 mmol) and $^t\text{BuNH}_2$ (0.5 mL) were added, and the reaction mixture was left stirring under refluxing condition. After completion of the reaction (24 h, TLC control), the reaction mixture was concentrated to a residue that on column chromatographic purification afforded amine **7** with some unidentified impurities. The impure amine **7** was used directly for the next step.

To a stirred solution of amine **7** in dry THF (5 mL) were added Et_3N (0.42 mL, 3 mmol) and DMAP (170 mg, 1 mmol) at room temperature. The resulting reaction mixture was cooled to 0 °C. $(\text{Boc})_2\text{O}$ (1.5 mmol) was then added to it. The stirring was continued for overnight without further cooling. After completion of the reaction, the reaction mixture was concentrated to a residue that on column chromatographic purification furnished the oxazolidinone **8** (59 mg, 0.15 mmol, 15% from **5**) with simultaneous Boc protection followed by oxazolidinone ring formation.

Eluent for column chromatography: EtOAc/hexane (1/2, v/v); $[\alpha]_D^{26} = -80.0$ (c 0.28, CHCl_3); $R_f = 0.33$ (1/3, EtOAc/hexane); IR (neat, cm^{-1}) 3342, 1740, 1219, 769; ^1H NMR (300 MHz, CDCl_3) δ 1.31 (s, 3H), 1.40 (s, 9H), 1.53 (s, 3H), 4.25–4.30 (m, 1H), 4.36–4.50 (m, 3H), 5.37–5.43 (m, 2H), 5.89–6.01 (m, 1H), 6.25 (dd, $J = 9.0, 15.9$ Hz, 1H), 6.52 (d, $J = 16.0$ Hz, 1H), 7.32–7.39 (m, 5H); ^{13}C NMR (75 MHz, CDCl_3) δ 25.7, 28.1, 28.6, 54.8, 60.9, 76.5, 76.8, 78.3, 110.1, 120.4, 126.0, 127.0, 129.1, 129.3, 134.5, 135.0, 135.7, 155.8; HRMS (ESI TOF (+)) m/z $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{22}\text{H}_{30}\text{NO}_4$ 372.2175, measured 372.2169.

Compound 10. To a stirred suspension of D -galactose **9** (5 g, 27.75 mmol) in dry acetone (200 mL) was added I_2 (1.5 g, 5.9 mmol) at room temperature, and the reaction was allowed to stir at same temperature. After 24 h, excess iodine was quenched with aqueous saturated solution of $\text{Na}_2\text{S}_2\text{O}_3$, and the resulting mixture was concentrated. The reaction mixture was then extracted with EtOAc (3 × 100 mL), and the combined organic layer was dried over Na_2SO_4 and concentrated under reduced pressure to a residue that on purification by column chromatography gave 1,2;3,4-di-*O*-isopropylidene- α - D -galactopyranoside **10** (5.77 g, 22.2 mmol, 80%).

Eluent for column chromatography: EtOAc/hexane (1/4, v/v); $[\alpha]_D^{28} = -55.3$ (c 0.77, CHCl_3); $R_f = 0.29$ (1/2, EtOAc/hexane); IR (neat, cm^{-1}) 3532, 3462, 2989, 2930, 1380, 1170; ^1H NMR (300 MHz, CDCl_3) δ 1.30 (s, 6H), 1.41 (s, 3H), 1.49 (s, 3H), 2.61 (s, 1H, OH), 3.69–3.82 (m, 3H), 4.22–4.31 (m, 2H), 4.56–4.59 (m, 1H), 5.52 (d, $J = 4.95$ Hz, 1H); ^{13}C NMR (50 MHz, CDCl_3) δ 24.6, 25.2, 26.2, 26.3, 62.4, 68.6, 70.9, 71.0, 71.7, 96.6, 109.0, 109.7; HRMS (ESI TOF (+)) m/z $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{12}\text{H}_{21}\text{O}_6$ 261.1338, measured 261.1328.

Compound 11. 1,2;3,4-di-*O*-isopropylidene- α - D -galactopyranoside **10** (1.0 g, 3.84 mmol), PPh_3 (1.51 g, 5.76 mmol), and imidazole (785

mg, 11.52 mmol) were taken in a round-bottom flask, and dry toluene (20 mL) was added to it. The reaction mixture was stirred at room temperature with I_2 (1.46 g, 5.76 mmol). The dark brown reaction mixture was then heated at 80 °C for 3 h. After completion of the reaction (TLC control), the reaction mixture was quenched with MeOH (2 mL), and the resulting solution was concentrated to a residue. A saturated aqueous solution of $\text{Na}_2\text{S}_2\text{O}_3$ was added to it. The stirring of the solution was continued until it became colorless. Afterward, it was extracted with DCM (2 × 25 mL), and the combined organic layer was dried over Na_2SO_4 and concentrated under reduced pressure to a residue that on column chromatographic purification afforded the iodide **11** (1.22 g, 3.3 mmol, 86%).

Eluent for column chromatography: EtOAc/hexane (1/4, v/v); $[\alpha]_D^{28} = -56.8$ (c 0.71, CHCl_3); $R_f = 0.66$ (1/3, EtOAc/hexane); IR (neat, cm^{-1}) 3333, 2979, 2906, 1380, 1258, 1069; ^1H NMR (300 MHz, CDCl_3) δ 1.26–1.27 (m, 6H), 1.36 (s, 3H), 1.46 (s, 3H), 3.10–3.14 (m, 1H), 3.21–3.26 (m, 1H), 3.86–3.89 (m, 1H), 4.22–4.23 (m, 1H), 4.31–4.34 (m, 1H), 4.53–4.55 (m, 1H); ^{13}C NMR (75 MHz, CDCl_3) δ 8.6, 30.8, 31.2, 32.25, 32.3, 75.2, 82.9, 83.4, 83.8, 103.0, 115.1, 115.8; HRMS (ESI TOF (+)) m/z $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{12}\text{H}_{20}\text{IO}_5$ 371.0355, measured 371.0345.

General Procedure for Preparation of Compound 12. To a magnetically stirred suspension of Zn dust (680 mg, 10.4 mmol) and NH_4Cl (560 g, 10.38 mmol) in dry methanol (20 mL) was added cyanocobalamin (7 mg, 0.005 mmol). The stirring was allowed to continue for another 10 min. After that, a solution of **11** (370 mg, 1 mmol) in dry methanol (5 mL) was added, and the resulting solution was further stirred for 15 min. The reaction mixture was filtered through a Celite bed. The filtrate was evaporated under reduced pressure. The residue obtained was dissolved in ethyl acetate (30 mL) and washed with a mixture of brine and water (1:1 v/v, 10 mL each). The organic layer was dried over Na_2SO_4 and concentrated to give the hemiacetal **12** as a mixture of α and β anomer (220 mg, 0.9 mmol, 90%), which was directly used for the next step.

Compound 8 from Compound 12. To a stirred solution of hemiacetal **12** (245 mg, 1 mmol) in EtOH (6 mL) were added *trans*-2-phenylvinyl boronic acid (149 mg, 1 mmol) and $^t\text{BuNH}_2$ (0.5 mL), and the reaction mixture was left stirring. After completion of the reaction (24 h, TLC control), the reaction mixture was concentrated to a residue that on column chromatographic purification afforded amine **7** with some unidentified impurities.

To a stirred solution of amine **7** in THF (5 mL) were added Et_3N (0.42 mL, 3 mmol) and DMAP (170 mg, 1 mmol), and the resulting solution was cooled to 0 °C. $(\text{Boc})_2\text{O}$ (1.5 mmol) was added to it, and the stirring was continued for 12 h. Afterward, the reaction mixture was concentrated under reduced pressure to a residue that on column chromatographic purification furnished the oxazolidinone **8** (155 mg, 0.4 mmol, 40% from **12**).

Compound 13. To a 100 mL two necked oven-dried round-bottom flask fitted with a reflux condenser and septum was added Grubbs second generation catalyst (22 mg, 0.026 mmol) under argon atmosphere. Dry degassed CH_2Cl_2 (25 mL) was added to it through a syringe, and the resulting solution was left stirring. Compound **8** (100 mg, 0.26 mmol) in DCM (5 mL) was added through a syringe to the stirring solution. The septum was replaced with a glass stopper while the stirring was continued. The solution was refluxed for 24 h. The temperature of the mixture was cooled slowly to room temperature. The organic solvent was evaporated under reduced pressure to give a black residue that on column chromatographic purification gave **13** as a semisolid compound (37 mg, 0.133 mmol, 51%).

Eluent for column chromatography: EtOAc/hexane (1/9, v/v); $[\alpha]_D^{28} = +124.2$ (c 0.125, CHCl_3); $R_f = 0.29$ (1/3, EtOAc/hexane); IR (neat, cm^{-1}) 3026, 2791, 1745, 1216, 766; ^1H NMR (300 MHz, CDCl_3) δ 1.36–1.37 (m, 6H), 1.46 (s, 9H), 4.23 (d, $J = 5.9$ Hz, 1H), 4.57 (brm, 2H), 4.74 (d, $J = 6.6$ Hz, 1H), 5.72–5.84 (m, 2H); ^{13}C NMR (75 MHz, CDCl_3) δ 27.0, 28.3, 28.9, 51.7, 54.2, 69.7, 70.9, 71.4, 110.3, 123.3, 129.0, 155.4; HRMS (ESI TOF (+)) m/z $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{14}\text{H}_{22}\text{NO}_4$ 268.1549, measured 268.1544.

Compound 14. To a 50 mL round-bottom flask compound **13** (30 mg, 0.11 mmol) was taken in DCM (4 mL), and TFA (1 mL) was

added to it at 0 °C. The mixture was allowed to stir without further cooling. After 2 h, it was cooled to 0 °C, and water (1 mL) was added to it. The resulting solution was further left stirring for 1 h. After that, the reaction mixture was concentrated and coevaporated with toluene under reduced pressure to get a residue that on column chromatographic purification gave the diol **14** (22 mg, 0.095 mmol, 85%).

Eluent for column chromatography: EtOAc/hexane (2/1, v/v); $[\alpha]_D^{28} = +78.5$ (c 0.23, CHCl₃); $R_f = 0.30$ (4/1, EtOAc/hexane); IR (neat, cm⁻¹) 3211, 3140, 1740, 1654, 1261, 804; ¹H NMR (300 MHz, CDCl₃) δ 1.37 (s, 9H), 4.24–4.29 (m, 2H), 4.37 (brm, 1H), 4.57–4.60 (m, 1H), 5.75 (m, 2H); ¹³C NMR (50 MHz, CDCl₃) δ 28.7, 52.6, 54.4, 64.2, 67.5, 73.5, 124.8, 131.7, 156.6; HRMS (ESI TOF (+)) *m/z* [M + H]⁺ calcd for C₁₁H₁₈NO₄ 228.1236, measured 228.1274.

Compound 19. To a stirred suspension of D-mannose **17** (5 g, 27.75 mmol) in dry acetone (200 mL) was added I₂ (1.5 g, 5.9 mmol), and the reaction was allowed to stir at room temperature. After 24 h, excess iodine was quenched with an aqueous saturated solution of Na₂S₂O₃, and the resulting mixture was concentrated under vacuum. The aqueous portion was extracted with EtOAc (3 × 100 mL), and the combined organic layer was dried over Na₂SO₄ and concentrated under reduced pressure to a residue that on purification by column chromatography gave 1,2;5,6-di-O-isopropylidene-α-D-mannofuranoside **18** (5.78 g, 22.2 mmol, 80%).

To a 250 mL two necked dry round-bottom flask were added methyltriphenylphosphonium bromide (Ph₃PCH₃Br) (39.6 g, 111 mmol) and ^tBuOK (9.95 g, 88.8 mmol) under nitrogen atmosphere. Dry THF (100 mL) was added at 0 °C, and the stirring was continued without further cooling. After 1 h, the reaction mixture was again cooled to 0 °C, and the compound **18** (5.78 g, 22.2 mmol) in THF (25 mL) was added to it through a syringe. The solution was stirred until the reaction was completed (TLC control, 4 h). Afterward, the reaction mixture was quenched with an aqueous solution of NH₄Cl (25 mL). It was extracted with EtOAc (3 × 50 mL), and the combined organic layer was evaporated under reduced pressure to give a residue that on column chromatographic purification afforded the olefin **19** as a clear oil (4.87 g, 18.87 mmol, 85%).

Eluent for column chromatography: EtOAc/hexane (1/13, v/v); $[\alpha]_D^{28} = -35.7$ (c 1.0, CHCl₃); $R_f = 0.54$ (1/4, EtOAc/hexane); IR (neat, cm⁻¹) 3235, 1634, 1217, 764; ¹H NMR (300 MHz, CDCl₃) δ 1.29 (s, 3H), 1.33 (s, 3H), 1.35 (s, 3H), 1.48 (s, 3H), 2.21 (d, *J* = 8.0 Hz, 1H), 3.41 (dd, *J* = 6.7, 7.7 Hz, 1H), 3.91–4.05 (m, 3H), 4.33 (dd, *J* = 1.09, 7.44 Hz, 1H), 4.64 (t, *J* = 7.5 Hz, 1H), 5.25–5.37 (m, 2H), 5.99–6.10 (m, 1H); ¹³C NMR (50 MHz, CDCl₃) δ 24.8, 25.6, 26.9, 27.0, 67.4, 70.8, 76.3, 77.0, 79.4, 108.9, 109.5, 119.7, 134.6; HRMS (DART TOF (+)) *m/z* [M + H]⁺ calcd for C₁₃H₂₃O₅ 259.1545, measured 259.1542.

Compound 20. To a stirred solution of compound **19** (260 mg, 1 mmol) in DMF (2 mL) were added imidazole (275 mg, 4 mmol) and TBDPSCl (412 mg, 0.39 mL, 1.5 mmol), and the reaction mixture was allowed to stir for 24 h. Water (10 mL) was added to it, and the entire solution was extracted with EtOAc (2 × 10 mL). The combined organic layer was evaporated under reduced pressure to give a residue that on purification by column chromatography afforded the silyl ether **20** as clear oil (432 mg, 0.87 mmol, 87%).

Eluent for column chromatography: EtOAc/hexane (1/30, v/v); $[\alpha]_D^{28} = +37.2$ (c 1.47, CHCl₃); $R_f = 0.5$ (1/6, EtOAc/hexane); IR (neat, cm⁻¹) 2938, 2796, 1374, 1217, 765; ¹H NMR (300 MHz, CDCl₃) δ 1.07 (s, 9H), 1.11 (s, 3H), 1.14 (s, 3H), 1.31 (s, 3H), 1.32 (s, 3H), 3.82 (t, *J* = 7.2 Hz, 1H), 3.97–4.10 (m, 4H), 4.40–4.44 (m, 1H), 5.02–5.16 (m, 2H), 5.42–5.54 (m, 1H), 7.35–7.42 (m, 6H), 7.71–7.76 (m, 4H); ¹³C NMR (50 MHz, CDCl₃) δ 20.1, 25.3, 25.5, 26.4, 27.4, 28.0, 67.5, 72.8, 77.2, 79.1, 79.9, 108.1, 109.4, 118.5, 127.3, 127.6, 129.4, 129.7, 134.1, 134.8, 135.1, 136.4, 136.6; HRMS (ESI TOF (+)) *m/z* [M + Na]⁺ calcd for C₂₉H₄₀O₅SiNa 519.2537, measured 519.2543.

Compound 21. To a 100 mL round-bottom flask was added compound **20** (500 mg, 1 mmol) dissolved in 80% AcOH (10 mL), and the solution was allowed to stir at room temperature. After 6 h, the reaction mixture was concentrated and coevaporated with toluene under reduced pressure to obtain an oily residue that on column

chromatographic purification furnished diol **21** (342 mg, 0.75 mmol, 75%).

Eluent for column chromatography: EtOAc/hexane (1/2, v/v); $[\alpha]_D^{28} = +19.4$ (c 0.88, CHCl₃); $R_f = 0.56$ (2/5, EtOAc/hexane); IR (neat, cm⁻¹) 3033, 2961, 1641, 1216, 762; ¹H NMR (300 MHz, CDCl₃) δ 1.09 (s, 9H), 1.15 (s, 3H), 1.25 (s, 3H), 2.36 (s, 1H, OH), 2.71 (s, 1H, OH), 3.67–3.78 (m, 3H), 3.93 (dd, *J* = 3.3, 7.9 Hz, 1H), 4.17–4.22 (m, 1H), 4.35–4.39 (m, 1H), 5.06 (d, *J* = 10.1 Hz, 1H), 5.15 (d, *J* = 16.9 Hz, 1H), 5.47–5.59 (m, 1H), 7.36–7.44 (m, 6H), 7.70–7.75 (m, 4H); ¹³C NMR (50 MHz, CDCl₃) δ 20.0, 25.5, 27.5, 27.9, 63.7, 72.8, 74.3, 78.9, 79.3, 108.8, 119.6, 127.5, 127.9, 129.8, 130.1, 133.7, 134.0, 134.5, 136.5; HRMS (ESI TOF (+)) *m/z* [M + Na]⁺ calcd for C₂₆H₃₆O₅SiNa 479.2224, measured 479.2231.

Compound 25. To the diol **21** (500 mg, 1.1 mmol) dissolved in THF/H₂O (9:1, 15 mL) was added NaIO₄ (320 mg, 1.5 mmol) at 0 °C, and the mixture was stirred for 1.5 h without further cooling. The reaction mixture was quenched with saturated Na₂S₂O₃ solution (5 mL) at 0 °C and extracted with EtOAc (2 × 10 mL). The combined organic layer was dried over Na₂SO₄ and concentrated under reduced pressure to give the crude aldehyde **22**, which was immediately used for the next step.

To a stirred solution of aldehyde **22** in EtOH (10 mL) was added TBAF (1.3 mL, 1 M solution in THF), and the reaction mixture was left stirring at room temperature. After 30 min, *trans*-2-phenylvinyl boronic acid (165 mg, 1.1 mmol) and ^tBuNH₂ (0.5 mL) were added to the reaction mixture, and it was refluxed for 24 h. Solvent was removed under reduced pressure, and the resulting oil was purified by flash chromatography to yield amino alcohol **24** with some unidentified impurities (TLC).

To the stirring solution of amine **24** in dry THF (5 mL) were added Et₃N (0.42 mL, 3 mmol) and DMAP (170 mg, 1 mmol) at room temperature. The resulting mixture was cooled to 0 °C. (Boc)₂O (1.5 mmol) was added to it, and the mixture was stirred overnight without further cooling. After 12 h, the reaction mixture was concentrated to a residue that on column chromatographic purification furnished the oxazolidinone **25** (89 mg, 0.23 mmol, 21% from **21**).

Eluent for column chromatography: EtOAc/hexane (1/9, v/v); $[\alpha]_D^{26} = +66.0$ (c 0.24, CHCl₃); $R_f = 0.3$ (1/3, EtOAc/hexane); IR (neat, cm⁻¹) 3026, 2761, 1745, 1216, 770; ¹H NMR (300 MHz, CDCl₃) δ 1.31 (s, 3H), 1.40 (s, 9H), 1.53 (s, 3H), 4.25–4.30 (m, 1H), 4.35–4.49 (m, 3H), 5.37–5.43 (m, 2H), 5.89–6.01 (m, 1H), 6.25 (dd, *J* = 9.1, 16.0 Hz, 1H), 6.52 (d, *J* = 16.0 Hz, 1H), 7.32–7.39 (m, 5H); ¹³C NMR (50 MHz, CDCl₃) δ 25.8, 28.1, 28.6, 54.8, 60.9, 76.6, 76.8, 78.4, 110.1, 120.4, 126.0, 127.0, 129.1, 129.3, 134.5, 135.0, 135.7, 155.8; HRMS (ESI TOF (+)) *m/z* [M + H]⁺ calcd for C₂₂H₃₀NO₄ 372.2175, measured 372.2167.

Compound 26. To a 50 mL two necked oven-dried round-bottom flask fitted with a reflux condenser and septum was added Grubbs second generation catalyst (11 mg, 0.013 mmol, 10 mol %) under argon atmosphere. Dry degassed CH₂Cl₂ (15 mL) was added to it, and the reaction mixture was left stirring. Compound **25** (50 mg, 0.13 mmol) in DCM (5 mL) was added through a syringe to the stirring solution. The septum was replaced with a glass stopper, and the solution was refluxed with stirring for 24 h. The temperature of the mixture was cooled slowly to room temperature. The organic solvent was evaporated under reduced pressure to give a black residue that on column chromatographic purification afforded **26** as a semisolid compound (17 mg, 0.062 mmol, 48%).

Eluent for column chromatography: EtOAc/hexane (1/9, v/v); $[\alpha]_D^{28} = -134.0$ (c 0.112, CHCl₃); $R_f = 0.3$ (1/3, EtOAc/hexane); IR (neat, cm⁻¹) 3069, 1745, 1217, 765; ¹H NMR (300 MHz, CDCl₃) δ 1.36–1.38 (m, 6H), 1.46 (s, 9H), 4.24 (d, *J* = 5.4 Hz, 1H), 4.57 (brm, 2H), 4.75 (d, *J* = 6.3 Hz, 1H), 5.72–5.84 (m, 2H); ¹³C NMR (75 MHz, CDCl₃) δ 27.0, 28.3, 28.9, 51.7, 54.2, 69.7, 70.8, 71.4, 110.3, 123.3, 129.0, 155.4; HRMS (ESI TOF (+)) *m/z* [M + H]⁺ calcd for C₁₄H₂₂NO₄ 268.1549, measured 268.1545.

Compound 27. To a solution of compound **26** (30 mg, 0.11 mmol) in DCM (4 mL) was added TFA (1 mL) at 0 °C, and the reaction was allowed to stir without further cooling. After 2 h, the reaction mixture was again cooled to 0 °C, and water (1 mL) was

added. The resulting solution was left stirring for another 1 h. After that, the reaction mixture was concentrated and coevaporated with toluene under reduced pressure to obtain a residue that on column chromatographic purification afforded the diol **27** (23 mg, 0.1 mmol, 91%).

Eluent for column chromatography: EtOAc/hexane (2/1, v/v); $[\alpha]_D^{28} = -73.7$ (*c* 0.16, CHCl₃); $R_f = 0.29$ (4/1, EtOAc/hexane); IR (neat, cm⁻¹) 3420, 2926, 1723, 1219, 768; ¹H NMR (300 MHz, CDCl₃) δ 1.42 (s, 9H), 3.65 (brs, 1H), 4.08 (brs, 1H), 4.29–4.32 (m, 2H), 4.40 (brm, 1H), 4.63 (dd, *J* = 4.2, 5.8 Hz, 1H), 5.79 (brm, 2H); ¹³C NMR (50 MHz, CDCl₃) δ 28.7, 52.6, 54.3, 64.2, 67.3, 73.6, 124.6, 131.8, 156.6; HRMS (ESI TOF (+)) *m/z* [M + H]⁺ calcd for C₁₁H₁₈NO₄ 228.1236, measured 228.1231.

■ ASSOCIATED CONTENT

● Supporting Information

General experimental details, full characterization and copies of ¹H NMR and ¹³C NMR spectra of compounds **2**, **3**, **5**, **8**, **10**, **11**, **13**, **14**, **19**, **20**, **21**, **25**, **26**, and **27** are provided. This material is available free of charge via the Internet at <http://pubs.acs.org>.

■ AUTHOR INFORMATION

Corresponding Author

*E-mail: akshaw55@yahoo.com.

Notes

The authors declare no competing financial interest.

■ ACKNOWLEDGMENTS

We are thankful to Sophisticated Analytical Instrument Facility, CDRI for providing spectral data, DST (Project SR/SI/OC-17/2010) for financial support and Mr. A.K. Pandey for technical assistance. P. G. thanks CSIR for awarding Senior Research Fellowship. CDRI communication no. 8302.

■ REFERENCES

- (1) Chen, X.; Fan, Y.; Zheng, Y.; Shen, Y. *Chem. Rev.* **2003**, *103*, 1955–1977.
- (2) Mahmud, T. *Nat. Prod. Rep.* **2003**, *20*, 137–166.
- (3) Kameda, Y.; Asano, N.; Teranoshi, M.; Amatsui, K. *J. Antibiot.* **1980**, *33*, 1573–1574.
- (4) Hudlicky, T.; Tian, X.; Koenigsberger, K.; Maurya, R.; Rouden, J.; Fan, B. *J. Am. Chem. Soc.* **1996**, *118*, 10752–10768 and references cited therein.
- (5) Hudlicky, T.; Olivo, H. F. *J. Am. Chem. Soc.* **1992**, *114*, 9694–9696.
- (6) Elango, S.; Yan, T.-H. *J. Org. Chem.* **2002**, *67*, 6954–6959.
- (7) Yamada, K.-I.; Yamashita, M.; Sumiyoshi, T.; Nishimura, K.; Tomoika, K. *Org. Lett.* **2009**, *11*, 1631–1633 and references cited therein.
- (8) Johnson, C. R.; Golebiowski, A.; Sundram, H.; Miller, M. W.; Dwiahy, R. L. *Tetrahedron Lett.* **1995**, *36*, 653–654 and references cited therein.
- (9) Pitzer, K.; Hudlicky, T. *Synlett.* **1995**, 803–805 and references cited therein.
- (10) Hudlicky, T.; Nugent, T.; Griffeth, W. *J. Org. Chem.* **1994**, *59*, 7944–7946.
- (11) Tian, X.; Maurya, R.; Königsberger, K.; Hudlicky, T. *Synlett* **1995**, 1125–1128 and references cited therein.
- (12) Paulsen, H.; Roben, W.; Heiker, F. R. *Chem. Ber.* **1981**, *114*, 3242–3253.
- (13) Hudlicky, T.; Olivo, H. F. *Tetrahedron Lett.* **1991**, *32*, 6077–6080.
- (14) Toung, R. L.; Liu, Y.; Muchowski, J. M.; Wu, Y.-L. *Tetrahedron Lett.* **1994**, *35*, 1639–1642.
- (15) Trost, B. M.; Pulley, S. R. *Tetrahedron Lett.* **1995**, *36*, 4737–4740.
- (16) Angelaud, R.; Babot, O.; Charvat, T.; Landais, Y. *J. Org. Chem.* **1999**, *64*, 9613–9624.
- (17) Ovva, H.; Codée, J. D. C.; Lastdrager, B.; Overkleeft, H. S.; van der Marel, G. A.; van Boom, J. H. *Tetrahedron Lett.* **1999**, *40*, 5063–5066.
- (18) Spielvogel, D.; Kammerer, J.; Keller, M.; Prinzbach, H. *Tetrahedron Lett.* **2000**, *41*, 7863–7867.
- (19) Lysek, R.; Favre, S.; Vogel, P. *Tetrahedron* **2007**, *63*, 6558–6572.
- (20) Lysek, R.; Schütz, C.; Vogel, P. *Bioorg. Med. Chem. Lett.* **2005**, *15*, 3071–3075.
- (21) Alegret, C.; Benet-Buchholz, J.; Riera, A. *Org. Lett.* **2006**, *8*, 3069–3072.
- (22) Pandey, G.; Tiwari, K. N.; Puranik, V. G. *Org. Lett.* **2008**, *10*, 3611–3614.
- (23) Chang, Y.-K.; Lo, H.-J.; Yan, T.-H. *Org. Lett.* **2009**, *11*, 4278–4281.
- (24) Trost, B. M.; Malhotra, S.; Olson, D. E.; Maruniak, A.; Bois, J. D. *J. Am. Chem. Soc.* **2009**, *131*, 4190–4191.
- (25) Jana, C. K.; Grimme, S.; Studer, A. *Chem.—Eur. J.* **2009**, *15*, 9078–9084.
- (26) Chappell, D.; Drew, M. G. B.; Gibson, S.; Harwood, L. M.; Russell, A. T. *Synlett* **2010**, 517–520.
- (27) Lu, P.-H.; Yang, C.-S.; Devendar, B.; Liao, C.-C. *Org. Lett.* **2010**, *12*, 2642–2645.
- (28) Norsikian, S.; Soulé, J.-F.; Cannillo, A.; Guillot, R.; Dau, M.-E. T. H.; Beau, J.-M. *Org. Lett.* **2012**, *14*, 544–547.
- (29) Ducatti, D. R. B.; Massi, A.; Noseda, M. D.; Duarte, M. E. R.; Dondoni, A. *Org. Biomol. Chem.* **2009**, *7*, 576–588.
- (30) Ghosal, P.; Kumar, V.; Shaw, A. K. *Tetrahedron* **2010**, *66*, 7504–7509.
- (31) Ghosal, P.; Kumar, V.; Shaw, A. K. *Carbohydr. Res.* **2010**, *345*, 41–44.
- (32) Reddy, L. V. R.; Swamy, G. N.; Shaw, A. K. *Tetrahedron: Asymmetry* **2008**, *19*, 1372–1375.
- (33) Reddy, L. V. R.; Reddy, P. V.; Shaw, A. K. *Tetrahedron: Asymmetry* **2007**, *18*, 542–546.
- (34) Pal, P.; Shaw, A. K. *Tetrahedron* **2011**, *67*, 4036–4047.
- (35) Kumar, V.; Das, P.; Ghosal, P.; Shaw, A. K. *Tetrahedron* **2011**, *67*, 4539–4546.
- (36) Ghosal, P.; Sharma, D.; Kumar, B.; Meena, S.; Sinha, S.; Shaw, A. K. *Org. Biomol. Chem.* **2011**, *9*, 7372–7383.
- (37) Petasis, N. A.; Akritopoulou, I. *Tetrahedron Lett.* **1993**, *34*, 583–586.
- (38) Petasis, N. A.; Zavialov, I. A. *J. Am. Chem. Soc.* **1997**, *119*, 445–446.
- (39) Petasis, N. A.; Goodman, A.; Zavialov, I. A. *Tetrahedron* **1997**, *53*, 16463–16470.
- (40) Petasis, N. A.; Zavialov, I. A. *J. Am. Chem. Soc.* **1997**, *119*, 445–446.
- (41) Petasis, N. A.; Zavialov, I. A. *J. Am. Chem. Soc.* **1998**, *120*, 11798–11799.
- (42) (a) Kleban, M.; Kautz, U.; Greul, J.; Hilgers, P.; Kugler, R.; Dong, H.-Q.; Jäger, V. *Synthesis* **2000**, 1027–1033. (b) Bernet, B.; Vasella, A. *Helv. Chim. Acta* **1979**, *62*, 1990–2016.
- (43) Kartha, K. P. R. *Tetrahedron Lett.* **1986**, *27*, 3415–3416.
- (44) Kim, C.; Hoang, R.; Theodorakis, E. A. *Org. Lett.* **1999**, *1*, 1295–1297.